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Chapter 10 21 Prentice-Hall, Inc. Name: \_\_\_\_\_ Date: \_\_\_\_\_

Chemistry 10-3 Practice Problems (Continued) 13. Find the empirical formula of a compound, given that the compound is found to be 47.9 percent zinc and 52.1 percent chlorine by mass. 20. Find the molecular formula of a compound that contains 42.56-g of palladium and 0.80-g of hydrogen.

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10-3 Practice Problems. 1. Find the percentage composition of a

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7. A sample of a compound that has a mass compound that contains 1.94-g of of 0.432-g is analyzed. The sample is carbon, 0.48-g of hydrogen, and 2.58-g found to be made up of oxygen and of sulfur in a 5.00-g sample of the. fluorine only.

### 10-3 Practice Problems - Lake Stevens School District

10.2 Mole to Mole & Mole to Volume Relationships \*For all the problems on this page, first find the molar mass of the compound. 1. What is the mass of 9.45 mol of aluminum oxide? mol  $Al_2O_3$  + 3 (l A GO q 63,5ù? - 2. What is the mass of  $4.52 \times 10^{-3}$  mol of ethylbenzene,  $C_6H_5CH_2CH_3$ ? \*Ethylbenzene is a hydrocarbon that is produced by burning coal.

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Practice Problems: Solutions (Answer Key) What mass of solute is needed to prepare each of the following solutions? a. 1.00 L of 0.125 M  $K_2SO_4$  21.8 g  $K_2SO_4$  b. 375 mL of 0.015 M NaF 0.24 g NaF c. 500 mL of 0.350 M  $C_6H_{12}O_6$  31.5 g  $C_6H_{12}O_6$ ; Calculate the molarity of each of the following solutions:

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CHAPTER 10 SOLUTIONS MANUAL The MoleThe Mole Solutions Manual Chemistry: Matter and Change • Chapter 10 161 Section 10.1 Measuring Matter page 320–324 Practice Problems pages 323–324 1. Zinc (Zn) is used to form a corrosion-inhibiting surface on galvanized steel.

## **Chemistry Chapter 10 Practice Problems Answers**

10—2 Practice Problems (continued) 17 If you burned 6.10 x

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1024 molecules of ethane ( $C_2H_6$ ), what mass of ethane did you burn? (JAÖ cutg 18 Hovl many form units are in 5. Date Class 2Ä. A chemical reaction produces 0.37 mol of  $N_2$  gas. What volume will that gas er with a volum of 694 L 25. A c contains how many moles of oxygen at 'mol lq t 3R

### 10-2 Practice Problems WS - Ms. Bloedorn's Class

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### Useful Advice - Home

0.710 kg of sodium carbonate (washing soda),  $Na_2CO_3$ , in 10.0 kg of water—a saturated solution at  $0^\circ C$ ; 125 g of  $NH_4NO_3$  in 275 g of water—a mixture used to make an instant ice pack; 25 g of  $Cl_2$  in 125 g of dichloromethane,  $CH_2Cl_2$ ; 0.372 g of histamine,  $C_5H_9N$ , in 125 g of chloroform,  $CHCl_3$

### 8.3: Concentrations of Solutions (Problems) - Chemistry



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...

Ksp Problems - Chemistry Name: \_\_\_\_\_ 1) The value of  $K_{sp}$  of  $AgCl$  is  $1.8 \times 10^{-10}$ . What would be the molar concentration of  $Ag^+$  and  $Cl^-$  in pure water placed in contact with solid  $AgCl(s)$ ?  
 $K_{sp} = [Ag^+][Cl^-]$   $1.8 \times 10^{-10} = [x][x]$   $x = 0.000013416$  M or  $1.3416 \times 10^{-5}$  M 2) The molar solubility of  $PbCl$  ...

### **Ksp Problems - Chemistry**

3—3 Practice Problems 1. How many protons and electrons are 8. What is the name of element that has present in a vanadium atom? atoms that contain 25 protons? 2. How many protons and electrons are 9. What is the name of the element that has present in a nitrogen atom? atoms that contain 82 protons? 3. How many protons and electrons are 10.

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