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SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1.

c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released.

2. d The energy required to remove an

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electron from a neutral atom is the atom's

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The Periodic Table

Answers

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2.

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SECTION 1 SHORT
ANSWER Answer the

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following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design.

Modern Chemistry
Chapter 5 Review
The Periodic Law
Answer Key

CHAPTER 5 REVIEW.

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SECTION 5-3. SHORT ANSWER
Answer the following questions in the space provided. 1. When an electron is added to a neutral atom, energy is .
(a) always absorbed
(c) either absorbed or released.
(b) always released
(d) burned away. 2.

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Section 1

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SHORT ANSWER

Answer the following questions in the space provided. 1. Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53.

- What is its atomic number? 127.
- What is its mass number? atomic number.
- Is the element's position in a modern periodic table

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chxistina. Terms in this

set (36) The order of
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periodic table is based
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number of protons in
the nucleus the

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number of neutrons in
the nucleus

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Science Chapter 5 The Periodic Table. The atomic mass of an element is. Which statement is true about the metal.... At room temperature none of the metals.... Which general statement does NOT apply.... a weighted average of

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the masses of an element's isotopes. Silicon is a better conductor of electric current than sulfur....

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ANSWER Answer the
following questions in
the space provided. 1.
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Answers Section 3

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Answers 5 1

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Chapter 5 Review The Periodic Law Answers 5 1 the modern periodic law states that the

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physical and chemical properties of an element are functions of its atomic number the discovery of the noble gases changed

Review The Periodic Law Answer Sheet

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SECTION 5-3 SHORT

ANSWER Answer the following questions in the space provided 1 When an electron is added to a neutral

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required to remove an
electron from an atom
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ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design.

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Answers

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