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Writing And Naming Binary Compounds

There are four main rules to naming binary molecular compounds

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and four easy steps to
the naming process.
Rule 1: If both
elements in the
compound are in the
same period on the
periodic table, the...

**Naming & Writing
Formulas for Binary
Molecular
Compounds ...**

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Binary covalent compounds are compounds made up of only two elements, such as carbon dioxide. Prefixes are used in the names of binary compounds to indicate the number of atoms of each nonmetal present. The following table lists the most common prefixes for binary covalent compounds. In general, the prefix mono- is rarely used.

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How to Name Binary Covalent Compounds - dummies

WRITING AND NAMING
BINARY COMPOUNDS
WORKSHEET. WKS 7.1
- Beginning Naming &
Formula Writing (2
pages) Name the
following ionic
compounds. Be sure to
check for Roman
Numerals! Na_3P .
 MgBr_2 . Ag_2Se . AlCl_3 .
 Ca_3P_2 .

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**Part 1 Nomenclature
(Writing and Naming
Binary Ionic ...**

The order for names in a binary compound is

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first the cation, then the anion. Use the name of cation with a fixed oxidation state directly from the periodic table. The name of the anion will be made from the root of the element's name plus the suffix "-ide."
Example 1: Write the name of the following formula: H_2S .

ChemTeam: **Nomenclature-** **Binary Compounds**

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deca-. The rules for using the prefix system of nomenclature of binary compounds can be summarized as follows. Generally, the less-electronegative element is written first in the formula, though there are a few exceptions. Carbon is always first in a formula and hydrogen is after nitrogen in a formula such as NH_3 .

7.11: Naming Binary

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**Molecular
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Naming Covalent
Compounds Naming
Binary Ionic
Compounds Polyatomic
Ions Naming with
Polyatomic Ions
Naming with Roman
Numerals Formula
Writing Naming Acids.
Naming Binary Ionic
Compounds. Rules. 1.
The Cation (positive
ion) is named first, the

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Anion second. 2.

Monoatomic Cations

take the element

name. Na^{+} --> Sodium.

Ca^{2+} --> Calcium.

Naming Binary Ionic Compounds - kentchemistry.com

To name acids, the prefix hydro- is placed in front of the nonmetal modified to end with -ic. The state of acids is aqueous (aq) because acids are found in water. Some

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common binary acids include: $\text{HF (g)} =$ hydrogen fluoride \rightarrow $\text{HF (aq)} =$ hydrofluoric acid. $\text{HBr (g)} =$ hydrogen bromide \rightarrow $\text{HBr (aq)} =$ hydrobromic acid.

2.10: Naming Binary, Nonmetal Compounds - Chemistry LibreTexts

Compound Name
Formula Search »
Moles to Grams

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Calculator » Common
Compounds List »
Chemical Equation
Balancer » Complete
List of Acids »
Complete List of Bases
» Molar to Mass
Concentration
Converter » Molar Mass
Calculator » Cations,
Anions List » Dilution
Calculator » Molarity
Calculator » Compound
Prefixes » Water
Insoluble ...

Compound Name

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Binary ionic compounds are named by naming the two ions in the compound: cation and anion.

Before we learn to name binary ionic compounds, we need some guidelines for

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naming the ions themselves. Let's consider cations first. Metals tend to form cations, and main-group metals form cations with a charge equal to the group number.

Interactive Student Tutorial - Pearson Education

Rules for naming Type I binary compounds 1. The cation is always named first and the

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anion second. 2.

Naming Chemical Compounds & Writing Chemical Formulas

For binary ionic compounds (ionic compounds that contain only two types of elements), the compounds are named by writing the name of the cation first followed by the name of the anion. For example, KCl, an ionic compound

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that contains K^+ and Cl^- ions, is named potassium chloride. This is the currently selected item.

Naming ions and ionic compounds (video) | Khan Academy

Roman numeral should be used when naming which compounds.
Binary Compounds List
Binary Ionic Compounds List.

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**Writing And Naming
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